SHORT REPORTS

¹H AND ¹³C NMR ASSIGNMENTS OF ISOQUINOLINE ALKALOIDS*

KITTISAK LIKHITWITAYAWUID, a,b NIJSIRI RUANGRUNGSI, b AND GEOFFREY A. CORDELL. a,c

- ^a Program for Collaborative Research in the Pharmaceutical Sciences, Department of Medicinal Chemistry and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, IL 60612, USA.
- b Department of Pharmacognosy, Faculty of Pharmaceutical Sciences, Chulalongkorn University, Bangkok 10330, Thailand.
- Address for correspondence.
- *Part XXVI in the series of "Traditional Medicinal Plants of Thailand." For Part XXV, see ref. 1.

(Received April 3, 1993)

ABSTRACT

Use of two-dimensional NMR techniques, including the homonuclear COSY, NOESY, HETCOR and COLOC experiments, to unequivocally assign the ¹H and ¹³C NMR data of the isoquinoline alkaloids (-)-salutaridine, (-)-dicentrine and (-)-tetrahydropalmatine is demonstrated.

INTRODUCTION

The isoquinoline alkaloids constitute an important group of tyrosine-derived natural products, possessing a diversity of structural type and functionality, and an amazingly wide range of biological activity.² Pharmacological properties of these naturally occurring bases can be illustrated, for instance, by the immunosuppressive activity of sinomenine (1),³ an alkaloid structurally closely related to (-)-salutaridine (2), the anticancer potential of (-)-dicentrine (3),^{4,5} and the sedative effect of (-)-tetrahydropalmatine (4).⁶ These biologically significant compounds, being a promising source of natural product-based medicinal agents, have attracted interest from researchers in several disciplines. In the present investigation, compounds 2, 3, and 4 have been selected as representatives of the morphinandienone, aporphine, and tetrahydroprotoberberine alkaloids for detailed NMR studies. We demonstrate herein the utilitization of several NMR techniques, including the homonuclear COSY, NOESY, APT, HETCOR and COLOC, for the definitive NMR assignment of these alkaloids.

MATERIALS AND METHODS

Plant Material and Isolation: (-)-Salutaridine (2), (-)-dicentrine (3), and tetrahydropalmatine (4) were obtained from the tubers of Stephania pierrei Diels.^{4,5}

- (-)-Salutaridine, **2**. M.p. 191°C; $[\alpha]^{20}_{\rm D}$ -65° (c 0.4, EtOH); UV $\lambda_{\rm max}$ (MeOH) 208 (log ϵ 4.34), 240 (4.09), 279 (3.61) nm; IR $\nu_{\rm max}$ (film) 3349, 2934, 1669, 1643, 1485, 1283, 1213, 1067 cm⁻¹; ¹H and ¹³C NMR, see Table 1; MS EI m/z (rel. int.) 327 (M+, 100), 313 (11), 312 (34), 299 (41), 285 (10), 284 (44), 268 (14), 256 (11), 242 (14), 226 (11).
- (-)-Dicentrine, 3. M.p. 170° C; $[_]^{20}_{D}$ -60° (c 0.2, CHCl₃); UV λ_{max} (MeOH) 220 (log ϵ 4.57), 230 (4.52), 270 (4.12), 281 (4.25), 304 (4.28) nm; IR ν_{max} (KBr) 2928, 1605, 1520, 1458, 1271, 1100 cm⁻¹; ¹H and ¹³C NMR, see Table 2; MS EI m/z (rel. int.) 339 (M+, 70), 338 (100), 324 (10), 307 (11), 296 (26), 265 (11).
- (-)-Tetrahydropalmatine, **4**. M.p. 139° C; $[\alpha]^{20}_{D}$ -292° (c 0.2, EtOH); UV λ_{max} (MeOH) 208 (log ϵ 4.42), 227 (3.99), 282 (3.42) nm; IR ν_{max} (KBr) 3004, 2816, 1609, 1510, 1495, 1258, 1107, 1024 cm⁻¹; ¹H and ¹³C NMR, see Table 3; MS EI m/z (rel int.) 355 (M+, 100), 354 (64), 324 (18), 190 (19), 165 (18), 164 (42), 149 (44).

Instrumentation: Melting points were determined on a Kofler hot plate and are uncorrected. Optical rotations were measured with a Perkin-Elmer 241 polarimeter. UV spectra were obtained on a Beckman DU-7 spectrometer and IR spectra measured on a Nicolet MX-1 FT-IR interferometer. ¹H NMR, homonuclear COSY, ¹³C NMR, APT and HETCOR spectra were recorded in CDCl₃, with TMS as internal standard, employing a Varian XL-300 instrument. Standard Varian pulse sequences were used. The COLOC experiments were carried out at 75.6 MHz on a Varian XL-300 nmr spectrometer, using a relaxation delay 1.0 sec, acquisition time 0.112 sec, $\Delta_1 = 25$ msec, $\Delta_2 = 30$ msec. Mass spectra were obtained with a Varian MAT 112S instrument operating at 70 eV.

RESULTS AND DISCUSSION

The structure of (-)-salutaridine (2) (sinoacutine)⁷, an alkaloid known to be a key intermediate for the biosynthesis of sinomenine (1),⁸ has been well established through NMR studies and syntheses;⁹⁻¹¹ however, complete ¹H and ¹³C NMR assignments of 2 have not been reported. In this study, a combination of the homonuclear COSY and NOESY experiments was used to obtain unequivocal ¹H NMR assignments of 2 (Table 1). Distinction between H-1 (δ 6.62, d, J = 8.3 Hz) and H-2 (δ 6.71, d, J = 8.3 Hz) was based on the NOE contour displayed between H-2 and 3-OCH₃ (δ 3.85) in the NOESY spectrum. Similarly, the downfield proton resonance at δ 7.62 (s) was assigned to H-5 according to its NOE with 6-OCH₃ (δ 3.72), leaving the signal at δ 6.29 (s) to be assigned to H-8. From the COSY spectrum, the aliphatic methine proton at δ 3.66 (d, J = 5.1 Hz) could be assigned to H-9, and this was corroborated by the NOE between H-8 and H-9. Assignment of the H₂-10, H₂-15, and H₂-16 methylene protons was performed by examination of the corresponding geminal and vicinal couplings in the COSY spectrum. The NOESY spectrum

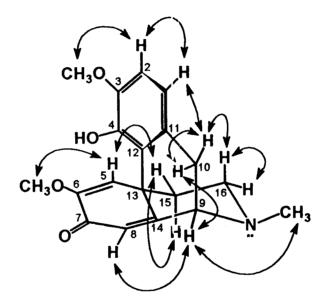


Fig.1. NOEs observed for (-)-saluaridine(2).

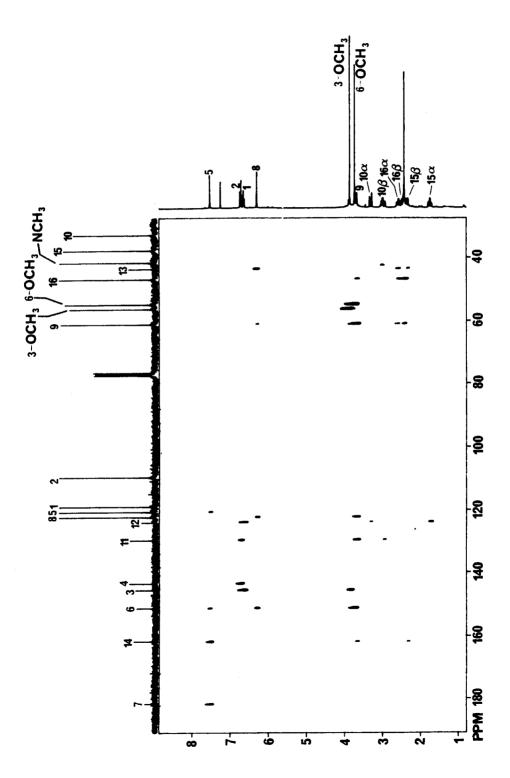


Fig.2. COLOC spectrum of (-)-salutaridine (2).

TABLE 1. ¹H and ¹³C NMR Assignments of (-)-Salutaridine (2)a.

Atom	1НР	13 C
1	6.62 (d, 8.3)	119.66
2	6.71 (d, 8.3)	110.36
3		146.22
4		144.19
5	7.52 (s)	121.37
6		151.82
7		182.27
8	6.29 (s)	123.04
9	3.66 (d, 5.1)	61.88
10α	3.30 (d, 17.1)	33.55
β	2.97 (dd, 17.1, 5.1)	
11		130.58
12		124.81
13		44.50
14		162.36
15	1.71 (ddd, 12.7, 12.7, 4.6)	38.53
	2.33 (ddd, 12.7, 2.7, 1.8)	
16	2.56 (ddd, 12.7, 4.6, 2.7)	47.82
_	2.45 (ddd, 12.7, 12.7, 1.8)	
3-OCH ₃	3.85 (s)	57.13
6-OCH ₃	3.72 (s)	55.66
NCH ₃	2.41 (s)	42.42

^a Recorded at 300 MHz for ^1H and 75.6 MHz for ^{13}C ; chemical shifts are reported in ppm (_) from TMS in CDCl₃.

^b Signal multiplicity and coupling constants are in parentheses.

TABLE 2. ¹H and ¹³C NMR Assignments of (-)-Dicentrine (3)^a

Atom	1Нр	13 C
1		141.70
1a		116.53
1b		126.45
2		146.53
3	6.48 (s)	106.68
За		126.61
4 ax.	3.07 (m)	29.19
eq.	2.58 (m)	
5 ax.	2.45 (m)	53.53
eq.	3.01 (m)	
ба		3.10 (m) 62.34
7 ax.	2.63 (m)	34.22
eq.	3.08 (m)	
. 7a		128.37
8	6.76 (s)	111.29
9		148.24
10		147.67
11	7.64 (s)	110.61
11a		123.54
OCH ₂ O	5.89 (d, 1.4)	100.51
2	6.04 (d, 1.4)	
9-OCH ₃	3.88 (s)	55.82
10-OCH ₃	3.89 (s)	56.06
NCH ₃	2.52 (s)	43.87

a Recorded at 300 MHz for ^1H and 75.6 MHz for ^{13}C ; chemical shifts are reported in ppm (_) from TMS in CDCl $_3$.

^b Signal multiplicity and coupling constants are in parentheses.

TABLE 3. ¹H and ¹³C NMR Assignments of (-)-Tetrahydropalmatine (4)^a

Atom	1Нр	13 C
1	6.70 (s)	108.27
2		147.13
3		147.13
4	6.58 (s)	111.03
4 a		126.51
5_	2.65 (m)	28.92
_	3.11 (m)	
<u>-</u> 6_	2.59 (m)	51.32
_	3.17 (m)	
8_	3.50 (d, 15.5)	53.80
_	4.21 (d, 15.5)	
8a		128.42
9		144.73
10		149.97
11	6.75 (d, 8.3)	110.61
12	6.84 (d, 8.3)	123.64
12a		127.45
13_	3.23 (dd, 15.5, 3.6)	36.16
_	2.79 (dd, 15.5, 11.5)	
14	3.49 (dd, 11.5, 3.6)	59.09
14a		129.43
2-OCH ₃	3.85 (s)	55.82
3-OCH ₃	3.83 (s)	55.60
9-OCH ₃	3.82 (s)	59.93
10-OCH ₃	3.80 (s)	55.60

^a Recorded at 300 MHz for ^{1}H and 75.6 MHz for ^{13}C ; chemical shifts are reported in ppm (δ) from TMS in CDCl₃.

^b Signal multiplicity and coupling constants are in parentheses.

also displayed NOE contours between H-5 and H-15 β , and H-9 and H-10 β , thereby confirming the assignments of these protons. In addition, steric proximity between H-1 and H-10_, and between H-10 β and H-16 β , was reflected by their respective NOE contours in the NOESY spectrum. A summary of NOEs of **2** is shown in Figure 1.

Unambiguous assignment of the ¹³C NMR resonances of 2 was accomplished by examination of the APT, HETCOR, and COLOC¹² spectra. The resonances of all of the protonated carbons were readily assigned through interpretation of the APT and HETCOR spectra. With the aid of the COLOC spectrum (Figure 2), all quaternary carbon signals could be unequivocally assigned. The resonance at δ 146.22 was assigned to C-3 based on its 3-bond coupling with H-1, whereas that at δ 144.19 was assigned to C-4 because of its 3-bond correlation with H-2. The assignment of C-3 was also supported by its 3-bond coupling with the 3-OCH₃ protons. On the other hand, the carbon at δ 151.82 displayed $^3J_{CH}$ correlation with the 6-OCH₃, and was therefore assigned to C-6. The resonance at δ 182.27 was assigned to C-7, and this assignment was supported by the 3-bond coupling between H-5 and C-7. The 3-bond coupling between the carbon at δ 130.58 and H-9 allowed this carbon to be assigned to C-11. The signal at δ 124.81 was assigned to C-12 from its 3-bond coupling with H-10_ and with H-15 α . The resonance at δ 44.50, showing $^3J_{CH}$ couplings with H-8 and H-16lpha, was assigned to C-13. The last quaternary carbon was assigned to C-14 according to its 3-bond couplings with H-5 and H-15β, and this completed the ¹³C NMR assignment of (-)-salutaridine (2) (Table 1).

The ¹H and ¹³C NMR assignments of (-)-dicentrine (**3**) and (-)-tetrahydropalmatine (**4**) were obtained in a similar manner. The results from analysis of the HETCOR spectrum in conjunction with the COLOC spectrum suggested a revision for the previous assignments of C-8 and C-11 of (-)-dicentrine (**3**)¹³ (Table 2). In the case of (-)-tetrahydropalmatine (**4**), the earlier assignments of C-4a, C-8a, C-12a, C-9, and C-10¹⁴ were revised (Table 3).

ACKNOWLEDGEMENTS

This work was supported, in part, by a grant from the Division of Cancer Treatment, National Cancer Institute, Bethesda, MD. We thank Dr. K. Zaw for the initial implementation of the COLOC pulse sequence at the College of Pharmacy, UIC. K.L. acknowledges the Graduate College of the University of Illinois at Chicago for a 1991-92 Deanûs Scholar Award.

REFERENCES

- 1. Guinaudeau, H., Lin, L.-Z., Ruangrungsi, N. and Cordell, G.A. (1993). J. Nat. Prod., submitted.
- 2. Shamma, M. (1972). The Isoquinoline Alkaloids: Chemistry and Pharmacology, Academic Press, New York, N.Y.
- 3. Hojo, H., Kondo, Y. Umeda, H. Tahira, T. and Hashimoto, Y. (1985). J. Immunopharmacol. 7, 33.
- 4. Likhitwitayawuid, K. (1992). Ph.D. Thesis, University of Illinois at Chicago, Chicago, I.L.
- 5. Likhitwitayawuid, K., Angerhofer, C.K., Chai, H., Pezzuto, J.M., Cordell, G.A. and Ruangrungsi, N. (1993). J. Nat. Prod., submitted.
- Preninger, V. (1975). The Alkaloids: Chemistry and Pharmacology, Academic Press, New York, N.Y., vol. 15, pp. 207-261.
- 7. Bentley, K.W. (1971). The Alkaloids: Chemistry and Pharmacology, Academic Press, New York, vol. 13, pp. 3-163.
- 8. Barton, D.H.R., Kirby, A.J. and Kirby, G.W. (1968). J. Chem. Soc. (C), 929.
- 9. Barnes, R.A. and Soeiro, O.M. (1981). Phytochemistry, 20, 543.
- 10. Kametani, T. Ihara, M. Fukumoto, K. and Yagi, H. (1969). J. Chem. Soc. (C), 2030.
- 11. Ludwig, W. and Schafer, H.J. (1986). Angew. Chem. Int. Ed. Engl. 11, 1025.
- 12. Kessler, H., Griesinger, C., Zarbock, J. and Loosli, H.R. (1984). J. Magn. Reson. 57, 331.
- 13. Shamma, M. and Moniot, J.L. (1978). Isoquinoline Alkaloids Research: 1972-1977, Plenum Press, New York, N.Y., p. 153.
- 14. Hughes, D.W. Holland, H.L. and Maclean, D.B. (1976). Can. J. Chem. 54, 2252.